

# Molecular Complexity from Polyunsaturated Substrates: The Gold Catalysis Approach

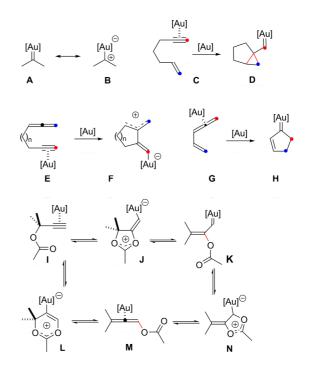
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# CONSPECTUS

ver the last two decades, electrophilic catalysis relying on platinum(II), gold(I), and gold(III) salts has emerged as a remarkable synthetic methodology. Chemists have discovered a large variety of organic transformations that convert a great assortment of highly functionalized precursors into valuable final products. In many cases, these methodologies offer unique features, allowing access to unprecedented molecular architectures. Due to the mild reaction conditions and high function compatibility, scientists have successfully developed applications in total synthesis of natural products, as well as in asymmetric catalysis. In addition, all these developments have been accompanied by the invention of well-tailored catalysts, so that a palette of different electrophilic agents is now commercially available or readily synthesized at the bench.

In some respects, researchers' interests in developing homogeneous gold catalysis can be compared with the Californian gold rush of the 19th century. It has attracted into its fervor thousands of scientists, providing a huge number of versatile and important reports. More notably, it is clear that the contribution to the art of organic



synthesis is very valuable, though the quest is not over yet. Because they rely on the intervention of previously unknown types of intermediates, new retrosynthetic disconnections are now possible.

In this Account, we discuss our efforts on the use of readily available polyunsaturated precursors, such as enynes, dienynes, allenynes, and allenenes to give access to highly original polycyclic structures in a single operation. These transformations transit via previously undescribed intermediates **A**, **B**, **D**, **F**, and **H** that will be encountered later on. All these intermediates have been determined by both ourselves and others by DFT calculations and in some cases have been confirmed on the basis of experimental data. In addition, dual gold activation can be at work in some of these transformations, for instance, from **E** to **F**.

Strikingly, we have found propargyl acetates to be particularly productive precursors. In a preliminary step upon electrophilic activation (complex I), they can lead to oxonium J or a vinylcarbenoid species K after 1,2-migration or complexed allenylester M from a formal 1,3-migration. All of them can serve as versatile entries for multievent processes. The propargyl cycle, sometimes called the golden carousel, involves species I - N), which lie in a close equilibrium. The control of this merry-go-round and its offshoots depends on the energy barriers associated with the subsequent reactions of these intermediates.

We illustrate these themes in this Account, focusing on the intriguing characteristics of gold catalysis.

## 1. Introduction

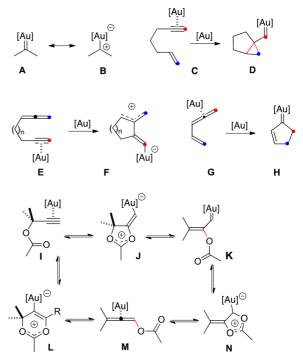
Consuming  $\pi$ -bonds to create  $\sigma$ -bonds is thermodynamically favorable. It is the foundation of a very vast number of synthetic transformations generally consisting of addition reactions. These addition processes to a  $\pi$ -component might involve highly reactive nucleophilic, electrophilic, or radical intermediates. They can also necessitate a catalytic activation of the  $\pi$  partner. Over the last two decades, we have concentrated some of our activity to the development of multiple  $\sigma$ -bonds formation sequences from readily available and highly modular precursors to reach molecular complexity in a one-pot operation. For that purpose, radical transformations have displayed a palette of valuable features. The addition of a radical intermediate to a  $\pi$  bond is generally an easy and fast process. It results in the generation of a new reactive radical intermediate, which can itself engage in a new homolytic process, and so on, yielding highly valuable cascade processes.<sup>1</sup> These can lead to polycyclic frameworks, including triquinanes,<sup>2</sup> as well heterocyclic platforms such as quinazolidones or guanidines,<sup>3</sup> as shown in Scheme 1.

While working out these radical approaches, we also focused our attention on the development of PtCl<sub>2</sub>-catalyzed cycloisomerizations,<sup>4</sup> which emerged in the 1990s with the seminal work of Chatani and Murai<sup>5</sup> and were then followed by highly inspiring studies from the groups of Fürstner<sup>6</sup> and Echavarren.<sup>7</sup> Since that initiation time, this chemistry has fully bloomed providing remarkable entries for molecular complexity.<sup>8</sup>

In a collaborative project, with the group of Jose Marco-Contelles, we initially examined the behavior of polyunsaturated dienyne precursors 1 and 3 (two alkenes and one alkyne) bearing an oxygenated group at the propargylic position. We immediately stumbled into a fascinating dichotomy of reactivity. When exposed to 5 mol % PtCl<sub>2</sub> in toluene at 80 °C, methoxyether 1, led in a single transformation creating four C-C bonds to a highly congested tetracyclic derivative **2** as a single diastereomer, which displays two fused cyclopropane moieties. In contrast, ester precursor 3 provided two products, 4 and 5, with an untouched olefin moiety showing the 1,2-acetate migration accompanied by the formation of a cyclopropyl group (Scheme 2).<sup>9</sup> Product 2 formally results from the transformation of the alkyne into two carbene intermediates, a reactivity reminiscent of previous work by Chatani and Murai.<sup>10</sup>

1,2-Ester migration products **4** and **5**, presumably involving the **I** to **N** golden carousel as calculated by Cavallo,<sup>11</sup>

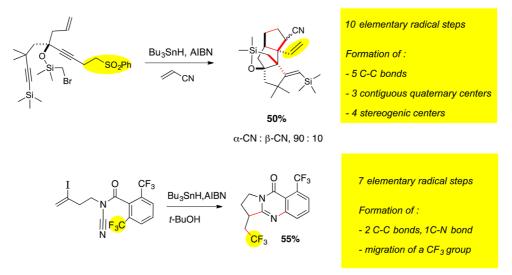
suggested a new and fecund type of transformations was available from this type of electrophilic catalysis. This proved to be completely confirmed.<sup>12</sup>



Additional examples of the ether–ester dichotomy were obtained from 5-en-1-yn-3-ol systems,<sup>13</sup> as well as in the transannular cycloisomerization of cycloundec-5-en-1-yne precursors (*p*-nitrobenzoate ester **7** and methoxy ether **9**), giving regioisomeric keto derivatives (Scheme 3).<sup>14</sup>

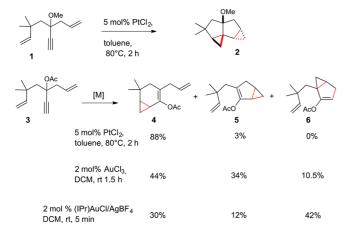
## 2. 1,2- or 1,3-Acetate Migration, a Versatile Entry to Molecular Complexity

2.1. Context. Interestingly, some intriguing, but barely quoted until the 2000s, literature precedents by Firmenich chemists Ohloff and Rautenstrauch showed that propargyl acetates could be easily activated with electrophilic metallic salts.<sup>15</sup> In 1984, Rautenstrauch published a very seminal work describing the Pd(II)-catalyzed formation of cyclopentenones from 1-ethynyl-2-propenyl acetates as in 14 to **15**.<sup>16</sup> This transformation is now coined as Rautenstrauch rearrangement or sometimes Ohloff-Rautenstrauch rearragement to refer to another seminal work by Ohloff<sup>17</sup> that describes the ZnCl<sub>2</sub>-mediated cycloisomerization of a 1,6enynyl acetate system 11 (Scheme 4). The common element of these transformations is the intervention of a formal 1,2-O-acyl migration. Soon, after the renaissance of this chemistry with PtCl<sub>2</sub> and inspiring reports by Ohe and Uemura with ruthenium,<sup>18</sup> it was rapidly established



#### SCHEME 1. Molecular Complexity via Radical Cascades That Consume $\pi$ -Bonds and Create $\sigma$ -Bonds

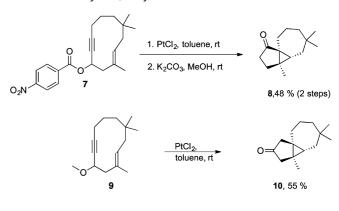
**SCHEME 2.** The Ether–Ester Dichotomy in the  $\ensuremath{\mathsf{PtCl}_2}\xspace$ -Catalyzed of Dienynes



by the groups of Toste<sup>19</sup> and Fürstner,<sup>20</sup> just to name a few, that gold catalysis was highly powerful for Rautenstrauchtype rearrangements and of course for a myriad of transformations.<sup>21</sup>

Related to our initial findings with PtCl<sub>2</sub>, and in collaboration with the group of Steven Nolan, we found that gold(I)– carbene complexes could alter the course of the reaction of **1** and favor a new rearrangement product **6**, already observed with AuCl<sub>3</sub> and ascribed to an increased stability of cationic intermediates brought by the NHC ligands on gold (Scheme 2).<sup>22,23</sup> Concomitant to these developments, it was shown by us that internal alkynes bearing an acetate group at the propargylic position undergo a 1,3-OAc migration upon catalysis with PtCl<sub>2</sub>, resulting in the formation of an allenyl ester derivative.<sup>24</sup> When flanked with a homoallyl

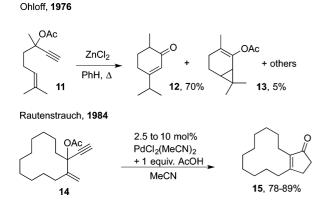
**SCHEME 3.** Regioisomeric Ketones from the Transannular Cycloisomerization of Macrocyclic 1,5-Enynes



group, the latter undergo a [3,3]-rearrangement leading to triene derivatives.<sup>25</sup>

The 1,3-OAc migration of propargyl acetates being used for decades with a variety of electrophilic metals  $(Ag^+, Cu^+, ...)$ <sup>26</sup> it was natural that gold electrophilic complexes could also catalyze this transformation as shown initially by Zhang.<sup>27</sup>

The mechanisms of the transformations of Scheme 2 have been at the center of intense debates. While the intervention of a platina- or an aura-cyclopropylcarbene intermediate of type **D** in the formation of **2** is probably admitted by everyone now on the basis of theoretical investigations and experimental evidence,<sup>7,28</sup> the formation of acetate migration products (**4**–**6**) is still under analysis, presumably involving the **I**–**N** manifold. It would take too much space of this Account to discuss all the pertinent literature. Suffice it to say that there is no obvious answer, especially when considering the results of Scheme 5.



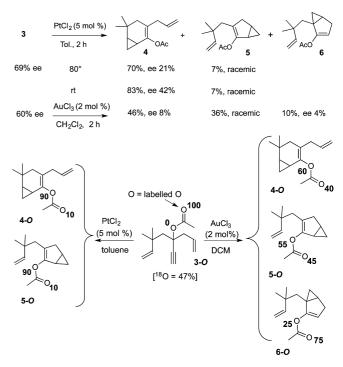
SCHEME 4. Seminal Works by Ohloff and Rautenstrauch

A partial chirality transfer is observed in the formation of the cyclohexene ring 4, while no ee was observed on the cyclopentene product **5**<sup>29</sup> Thus, the intervention of a free carbene intermediate as first suggested is unlikely. A more concerted pathway such as an initial cycloproponation stereodirected by the acetate group and its migration or a nucleophilic attack of the vinyl function on the oxonium intermediate is presumably involved, which corroborates chirality transfer studies by the Toste, Fürstner, and Fehr groups on shorter envne systems  $(1,4 \text{ and } 1,5)^{30}$  and calculations by de Lera<sup>31</sup> and Marco-Contelles and Soriano.32 Moreover, O-labeling analyses from 3-0 illustrate all the complexity of these transformations. The PtCl<sub>2</sub>-catalyzed process suggests a relatively clean 1,2-transfer while gold-catalyzed ones are consistent with an important scrambling of the oxygen atoms in the acetate migration step.<sup>29</sup>

**2.2. Toward Polycyclic Assemblies. 2.2.1. Cyclic Propargyl Acetates As Partners.** Besides applications in the total synthesis of natural products,<sup>29,33,34</sup> an attractive feature of these reactions is the opportunity to work out the straightforward preparation of polycyclic assemblies. Starting from easily accessible enynes **16** based on mixtures of diastereo-isomeric cyclic templates, a completely stereoconvergent<sup>35</sup> process catalyzed by platinum(II) or gold(I) and gold(III) affords a variety of tricyclic derivatives **17** accompanied by allenylesters **18** as byproducts.<sup>36</sup> Also remarkable is the possibility to run some of these reactions in ionic liquids<sup>37</sup> and to build medium sized rings even in neat conditions as shown in entry 6 of Table 1.

**2.2.2. Allenynes as Partners.** In the context of the development of this electrophilic catalysis, allenyne substrates have been quite useful for us, from a synthetic point of view and as mechanistic probes.<sup>38</sup> This is illustrated in Scheme 6, which shows two cases of subtle change in the catalyst structure that dramatically alters the fate of a reaction. Thus,

**SCHEME 5.** Mechanistic investigations in the 1,2-OAc migration products formation



allenyne 19 will very selectively give hydrindiene derivative 20 with halide containing gold and platinum catalysts, while halide free catalyst generally delivered Alder-ene products **21**.<sup>24,39</sup> Both type of products presumably originate from a nucleophilic attack of the allene on the electrophilically activated alkyne moiety as in E to F. Another striking example of contrasted behavior is given with the cycloisomerization of allenynol 22. Upon catalysis with PtCl<sub>4</sub>, a biscarbene reactivity as in the transformation of 9 into 10 (Scheme 3) is observed resulting from the activation of the alkyne partner and delivering fused bicyclic product 23, while with a cationic gold catalyst, electrophilic activation of the allene partner is observed leading to a nucleophilic attack of the alcohol function to provide ether **24**.<sup>40</sup> Because these processes are still under investigation from a mechanistic point of view in collaboration with mass spectroscopists Yves Gimbert (Université Joseph Fourier) and Denis Lesage (UPMC),<sup>41</sup> we have notably examined dual gold activation pathways.<sup>42</sup> However, we will not cover this work in this Account, and we will focus on allenylesters partners.

Indeed, allenyl esters, whether they originate from a direct 1,3-migration or a double 1,2-migration, have proven to be highly valuable partners that can be activated with platinum or gold catalysts or act as nucleophiles with a preactivated partner such as an alkyne.

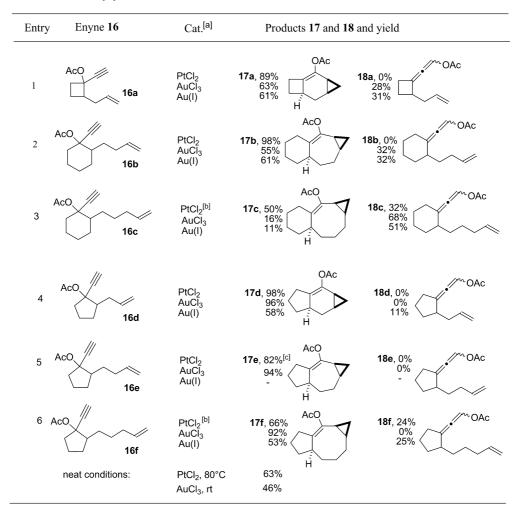


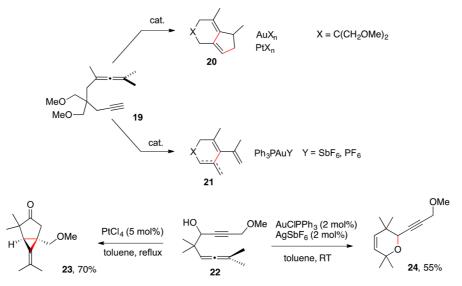
TABLE 1. Straightforward Access to Polycyclic Assemblies 17

<sup>*a*</sup>Au(l) refers to PPh<sub>3</sub>AuSbF<sub>6</sub> generated *in situ* from an equimolar mixture of AgSbF<sub>6</sub> and AuSIPPh<sub>3</sub>. Reactions were run at 80 °C in toluene with PtCl<sub>2</sub> (5 mol %) for 2 h and at rt in CH<sub>2</sub>Cl<sub>2</sub> with gold catalysts (2 mol %) for 5–10 min. <sup>*b*</sup>Reactions with PtCl<sub>2</sub> were run at rt for 24–48 h. <sup>*c*</sup>A minor diastereomer (<5%) was observed.

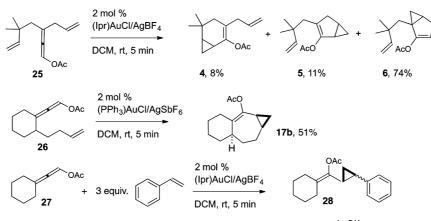
**2.2.3. Entering the Equilibrium from Allenyl Esters.** In a series of experiments, we could show that allenyl esters **25** and **26** are valuable precursors of previously described rearranged products **4**–**6** and **17b** as shown in Scheme 7. The ratio between **4** and **6** is altered, greatly favoring **6**, with the use of propargyl acetate **25** as the precursor (compare with Scheme 2). In some cases, with other cyclic templates, new rearrangement products were also observed. We could also engage allenyl ester **27** in an intermolecular cyclopropanation with styrene providing adduct **28** as a 3:1 mixture of diastereomers. All products, **4**–**6**, **17b**, and **28**, are further evidence of the **I**–**N** manifold. Other literature reports are also consistent with these findings.<sup>43</sup>

Our preliminary findings with allenene subtrates of type **29** showed that these substrates can be efficiently activated upon gold catalysis as in **0** to yield to carbene intermediate of type **H**, through a metallotropic Nazarov rerrangement. The carbene intermediate **H** has then the possibility to evolve through a hydrogen 1,2-migration delivering cyclopentadienyl substrates **30**<sup>44</sup> or to be trapped intramolecularly to give cyclopropyl moiety containing products 31 (Scheme 8).<sup>45</sup> Here also, it was judicious to apply a dual gold catalysis process starting from propargyl acetates to generate allenvlester precursors, in order to limit the number of steps to reach the reactive allenene substrate. We could thus use this principle for the expedient synthesis of polyquinane systems from propargyl acetate substrates, and it also culminated in the total synthesis of the linear triguinane natural product capnellene from easily accessible precursor 32. In a single reaction, using Echavarren's catalyst [Au1],<sup>46</sup> the tricyclic skeleton 33 of the natural product is assembled in high yield and completely stereoselectively. Following seven steps of





**SCHEME 7.** Electrophilic Activation of Allenyl Esters

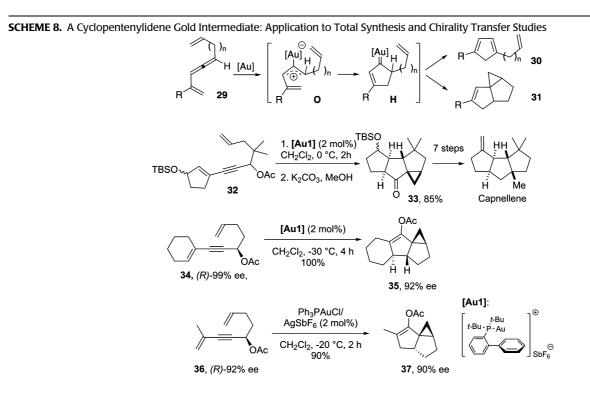




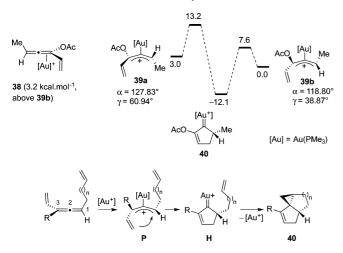
mainstream chemistry, the sesquiterpene natural product capnellene is obtained. Second, a simple enzymatic resolution of propargyl acetate substrates is possible furnishing precursors with a high enantiomeric enrichment. After a stereospecific 1,3migration, vinylallenyl esters precursors are indeed obtained with high ee's. An interesting effect of the degree of substitution of the allenene partner was evidenced.<sup>47</sup> While disubstituted vinylallenes undergo the previously mentioned rearrangement with no chirality transfer,<sup>44a</sup> trisubstitued allenenes stemming from precursors such as **34** and **36** provided the corresponding final products **35** and **37** with high to complete ee retention (Scheme 8).

To rationalize this intriguing observation, Vincent Gandon ran DFT calculations.<sup>45,47,48</sup> Coordination of gold to the central carbon of an allene may give rise to a so-called

"bent-allene" complex. An illustration of this is found in the modeling of the coordination of a cationic gold nucleus on trisubstituted vinylallenyl ester (Scheme 9). A minimum was found with twisted allylic cationic complex **39a**, compared with a diastereomeric form of it (complex **39b**) and  $\eta^2$ -allene complex **38**. Very interestingly, no planar structure could be modeled in that case. This marks a sharp contrast with disubstituted allene subtrates, which upon coordination with a cationic gold complex give birth to a fully delocalized allylic cation as minima. Thus, owing to allylic interactions, planar conformations of gold complexes of vinylallenes are disfavored compared with twisted ones. More importantly, the importance of bent allenes or twisted allylic cation intermediates lies in their ability to retain the chirality of the starting allene (Scheme 9). Indeed, inversion of



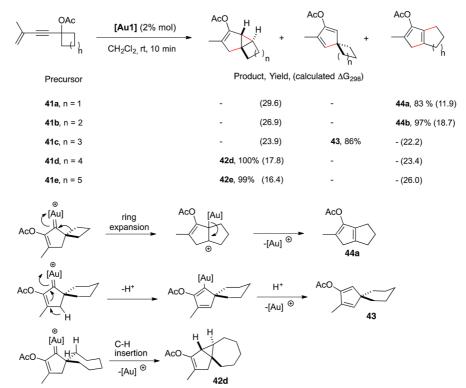
**SCHEME 9.** Bent Allenes or Twisted Allylic Cations as Intermediates

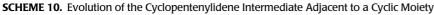


configuration of these species, and hence racemization of the substrate, would occur via planar transition states that are somehow difficult to reach if the starting allene is at least trisubstituted. In consequence, the metalla-Nazarov<sup>49</sup> pathway will take place with an attack of the vinyl group *anti* to the metal fragment on the configurationally stable intermediates **39b** and **P**, providing **40** and **H**, respectively. This theoretical insight is fully consistent with the experimental findings. Interestingly, we could also check that disubstituted allenes rapidly racemize in the reaction conditions.<sup>47</sup> Finally, it should be pointed out that transition metal complexes of tetraamino bent allenes have been isolated by Bertrand<sup>50</sup> and Fürstner<sup>51</sup> and that bent allene intermediates have also been proposed in related transformations by the groups of Toste<sup>52</sup> and Nevado.<sup>53</sup>

We examined the possible evolution of carbenes of type H in the absence of a pendant olefin on cyclic precursors of type 41 (Scheme 10). The ring size proved to be a crucial factor. Thus, cyclobutyl and cyclopentyl precursors 41a and 41b provided bicyclic fused cyclopentadienes 44a and 44b. These can be rationalized by a 1,2-carbocationic rearrangement releasing ring strain and leading to ring expansion. In sharp contrast, cycloheptyl and cyclooctyl templates 41d and 41e provided tricyclic cyclopropyl containing products 42d and 42e as single diastereomers. The latter presumably originate from a rare gold catalyzed 1,3 C-H insertion process,<sup>54,55</sup> presumably driven by entropic factors. Interestingly, with a six-membered ring, spiro product 43 has been obtained resulting from a prototropy sequence. In this case, no ring strain triggers the enthalpy demanding C-H insertion or the ring expansion leading to a cycloheptyl derivative, which is disfavored by both entropic and enthalpic factors. This proton shift was not possible to model; however DFT calculations were conducted on the ring expansion and ring fusion (C-H insertion) steps.

It is remarkable that the calculated energies of activation match perfectly the experimental findings. Notably, as n





increases, the energy to reach the ring fusion transition state decreases from 29.6 to 16.4 kcal·mol<sup>-1</sup>. Conversely, the ring expansion activation energies increase from 11.9 to 26 kcal·mol<sup>-1</sup>. For n = 3 (precursor **41c**), both processes require activation energies above 22 kcal·mol<sup>-1</sup>, favoring the elimination pathway. All these evolutions also illustrate a continuum between a carbocation and a carbene in the nature of intermediates of type **B** and **H**, which has been the matter of intense debate.<sup>56</sup>

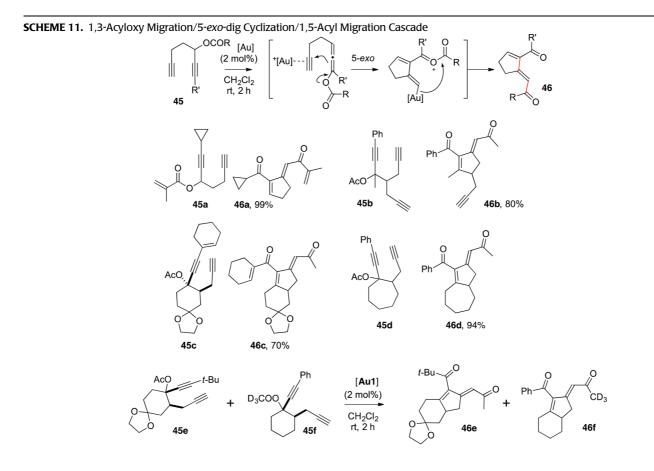
Unsymmetrical 1,6-diyne precursors **45** bearing an acetate group at the 3-propargylic position allowed confirmation of the higher reactivity of  $\eta^2$ -alkyne gold complexes toward nucleophilic attack compared with other unsaturated partners (alkenes, allenes). An interesting bonus in this sytem is the possibility to trap an acylium cation by a vinyl gold intermediate, resulting in an overall 1,3-acyloxy migration/ 5-*exo*-dig cyclization/1,5-acyl migration cascade, which delivers enones **46** (Scheme 11).<sup>57,58</sup>

The scope of this sequence proved to be wide as demonstrated by the efficient transformation of precursors 45a-d. The reaction of the labeled compound 45f, in the presence of 45e, did not lead to any deuterium scrambling on products 46e, f suggesting an intramolecular acylium shift. This was consistent with DFT calculations, which justified the Econfiguration of the exocyclic olefin.

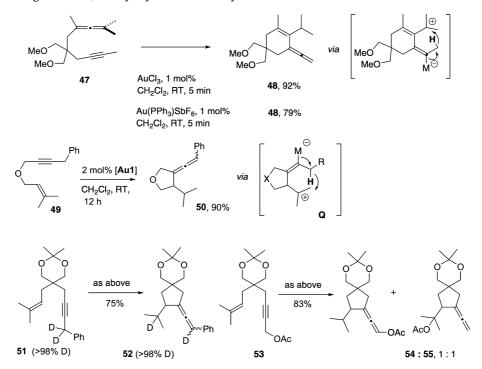
## 3. Other Opportunities for Molecular Complexities

3.1. Hydride Migration. 3.1.1. Allenyne Precursors. In the course of the exemplification of the platinum and gold catalyzed cycloisomerization of allenyne substrates, we encountered a very unique reactivity of precursors bearing a disubstituted alkyne moiety such as 47 (Scheme 12). An unprecedented formation of an exocyclic allene derivative (product **48**) was observed. Deuterium labeling studies<sup>22</sup> suggested a hydride migration from the methyl group to the allylic cation originating from the nucleophilic attack of the allene onto the activated alkyne partner as in **Q**. This 1,5hydride shift was also supported by DFT calculations. We could transpose this type of reactivity to envne systems bearing a prenyl group for which stabilization of a carbocation is fully operating. A series of exocyclic allenes has been synthesized (products **50**, **52**, and **54**).<sup>59</sup> In the case of propargyl acetates, a competing 1,5-acetate migration may take place giving allene 55.60 Synthetically valuable hydride transfers have also been disclosed by the group of Gagosz,<sup>61</sup> Helaja,<sup>62</sup> Hashmi,<sup>63</sup> and Alcaide and Almendros,<sup>64</sup> and they have been applied, for instance, to the preparation of allenes.<sup>65</sup>

**3.2. Alkyl Migration.** The gold-capped carbocationic intermediates of type **B** have been known to engage in different types of alkyl migration based rearrangements.



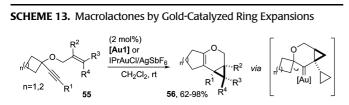
SCHEME 12. 1,5-Hydride Migration on 1,5-Prenylenynes To Provide Cyclic Allenes



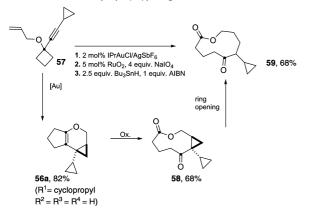
As shown in Scheme 10, ring strain release is a good driving force for this type of evolution. On the basis of this principle

and illustrative literature reports,<sup>66</sup> we could devise a three step synthetic sequence leading to macrocyclic lactones

from O-tethered 1,6-envnes containing a strained ring at the 3-position (Scheme 13).<sup>67</sup> Thus, cyclobutyl or cylopentyl precursors 55 upon cationic gold(I) catalysis reacted via intermediate ring expansion to efficiently provide fused tricyclic products 56. The latter easily undergo oxidative cleavage of the alkene function to provide ketolactone 58. Selective cyclopropyl ring-opening based on the intermediate



 $R^1$ = H, Ph,Me, *i*-Pr, *n*-Bu, vinyl, cyclopropyl, CO<sub>2</sub>Et ;  $R^2$  = H, Me ;  $R^3$  = H, Ph ;  $R^4$  = H, Me

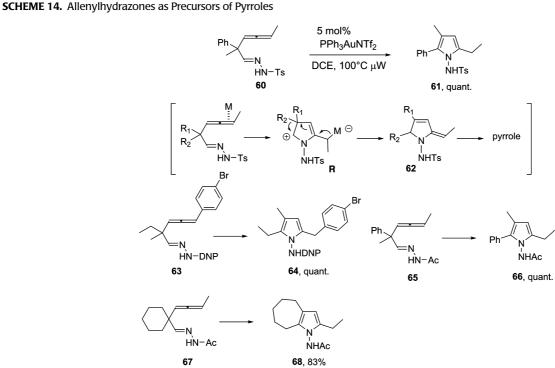


formation a O-stannyl ketyl radical allows the homologation of the macrocycle 59 to give 60.

Due to its high compatibility with heteroatoms and the mild reaction conditions required, gold catalysis is a valuable methodology for the preparation of various heterocyclic systems.<sup>68,69</sup> As illustrated before, allenes can be cleanly activated by cationic gold catalysts. We thought of taking advantage of this property with the use of nitrogen-based nucleophiles to provide a new synthetic access to pyrroles which remain interesting targets. We found that  $\beta$ -allenyl hydrazones were the proper substrates to undergo this transformation as shown by the full conversion tosylhydrazone precursor 60 in microwave conditions in the presence of Gagosz' catalyst (Scheme 14). Electrophilic activation of the allenyl moiety triggers the nucleophilic attack of the nitrogen atom. The resulting carbocationic center  $\alpha$  to the nitrogen atom (intermediate **R**) evolves through selective alkyl migration either ethyl vs methyl (as in product 64), phenyl vs methyl (as in product 66), or ring expansion (as in fused bicycle 68) to provide intermediate 62, which delivers the aromatic *N*-aminopyrrole system after prototropy.<sup>70</sup>

### 4. Concluding Remarks

In this Account, we have focused on the utilization of gold catalysts to produce highly functionalized and complex molecular scaffolds from readily available polyunsaturated





starting materials. Most of these reactions have shown high efficiency and selectivity. Their mechanisms are reasonably well-understood, notably thanks to DFT modeling. Applications in heterocylic chemistry, natural product synthesis, and asymmetric synthesis have been disclosed. Because substrate dependence is still very prominent in gold catalyzed transformation, some refinement in the control of the reactions is required. Essential progress in gold catalysis will be on ligand design in order to answer the frequent question: which catalyst for which reaction? More sustainable catalytic systems, which notably reduce the loading or allow recycling, are also highly desirable. Recent results along these lines such as the use polyoxometalate-based,<sup>71</sup> organometallic-platform-based,72 and NHC-cyclodextrinbased gold catalysts<sup>73</sup> have been published by us in the context of collaborative works with the teams of Bernold Hasenknopf, Hani Amouri, and Matthieu Sollogoub, respectively, at UPMC.

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#### **BIOGRAPHICAL INFORMATION**

**Louis Fensterbank** obtained his Ph.D. in 1993 from SUNY Stony Brook with Scott Sieburth. After a temporary lecturer position at UPMC in 1994, he was appointed Chargé de Recherche CNRS in 1995 in Max Malacria's team. In 2004, he obtained a professorship position at UPMC, and in 2008, he was nominated junior member of the Institut Universitaire de France.

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The authors declare no competing financial interest

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